

# PHYSICO – CHEMICAL, CORRELATION AND LINEAR REGRESSION ANALYSIS OF SMALL INDUSTRY AREA SOIL SAMPLES (SPRING SEASON)

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## 1. INTRODUCTION

Soil is one of the important and valuable resources in Nature. All living things are directly dependent on soil for their day to day needs and 95 % of human food is derived from the earth. Planning for healthy and productive soil is critical to human survival. Soil is a natural body made up of layers (soil horizons) of mineral constituents that differ from the parent materials in their morphological, physical, chemical and mineralogical properties. Soil is made up of broken rock particles that have been altered by chemical and mechanical processes such as weathering and erosion. Sewage sludge, industrial factory residues and intensive fertilisations are the most likely sources of soil, water and plant pollution. In many parts of the world, including India, the use of industrial wastewater in suburban areas is common practice.

## 2. PHYSICO CHEMICAL ANALYSIS OF SOIL SAMPLES

The physico-chemical parameters were studied for soil samples, such as physical properties and chemical properties. (color, taste, odor, total dissolved solids, total alkalinity, total hardness, calcium, magnesium, sodium, potassium, chloride, fluoride, sulphate, phosphate, nitrate, and carbonate). The relationship between the concentration of chemical compounds and elements present in the soil samples could be used to study the relationship between the soil parameters according to spring season. This study reveals that it creates the platform for many chemical compounds.

**Table - 2.1. Physico-chemical analysis of soil samples at spring season (SPS<sub>1</sub>-SPS<sub>15</sub>)**

Samples		Parameters															
		pH	EC	TDS	TA	TH	TUR	Ca <sup>2+</sup>	Mg <sup>2+</sup>	Na <sup>+</sup>	K <sup>+</sup>	Cl <sup>-</sup>	F <sup>-</sup>	SO <sub>4</sub> <sup>-</sup>	PO <sub>4</sub> <sup>-</sup>	NO <sub>3</sub> <sup>-</sup>	CO <sub>3</sub> <sup>-2-</sup>
Site-1	SPS <sub>1</sub>	8.6	1.19	797	224	122	1.2	22.08	12	50.04	5.00	258	1.5	140	4.6	2.0	166
	SPS <sub>2</sub>	8.7	1.26	844	220	128	1.4	22.06	12	41.08	5.06	230	0.5	145	5.8	2.4	168
	SPS <sub>3</sub>	8.4	1.24	830	226	132	1.5	22.01	12	44.07	5.09	234	1.5	180	4.8	2.8	158
Site-2	SPS <sub>4</sub>	8.3	1.25	837	224	126	1.4	21.06	12	41.02	4.02	254	1.5	185	5.0	2.9	140
	SPS <sub>5</sub>	8.5	1.20	804	222	136	1.6	22.05	12	50.01	4.01	232	1.0	150	4.9	2.5	130

	SPS <sub>6</sub>	8.7	1.21	810	214	120	1.0	19.06	10	40.01	3.06	236	0.5	70	5.8	2.0	160
Site-3	SPS <sub>7</sub>	8.8	1.18	790	230	128	1.1	20.04	12	40.05	2.05	268	1.0	65	5.9	3.0	152
	SPS <sub>8</sub>	8.6	1.23	824	232	134	1.5	22.07	12	39.07	2.01	214	1.0	70	6.2	2.0	162
	SPS <sub>9</sub>	8.0	1.05	703	238	122	1.3	20.03	13	50.05	3.03	242	1.5	60	5.0	3.1	142
Site-4	SPS <sub>10</sub>	8.5	1.26	844	234	132	1.9	20.01	12	51.00	3.09	230	0.5	55	4.8	2.0	176
	SPS <sub>11</sub>	8.4	1.22	817	232	138	1.7	21.03	11	50.06	3.07	220	1.0	70	4.6	1.5	160
	SPS <sub>12</sub>	8.6	1.21	810	234	134	1.3	21.06	11	40.07	2.07	246	1.5	75	5.2	1.6	172
Site-5	SPS <sub>13</sub>	8.7	1.22	817	236	126	1.2	21.08	11	40.04	2.06	272	1.5	160	5.8	2.4	168
	SPS <sub>14</sub>	8.6	1.21	810	232	128	1.1	20.06	12	51.05	2.00	260	0.5	165	5.9	3.1	144
	SPS <sub>15</sub>	8.0	1.20	804	230	122	1.0	20.01	12	51.08	2.01	256	1.0	160	4.7	2.4	148
<b>BIS (2012)</b>	6.5-8.5	300	500-2000	200-600	200-600	1-5	75-200	30-100	200-250	12	250-1000	1-1.5	200-400	1-5	150-350	200-600	

### 3. The correlation matrix studies for four seasons of soil samples by Physico-chemical method

The correlation matrix for physico-chemical parameters of soil samples for spring season of the industrial area samples (SPS<sub>1</sub>-SPS<sub>15</sub>) are given in Table 3.1.

**Table – 3.1. Correlation matrix for soil samples parameters in spring season at (SPS<sub>1</sub>-SPS<sub>15</sub>)**

Parameters	pH	EC	TDS	TA	TH	TUR	Ca <sup>2+</sup>	Mg <sup>2+</sup>	Na <sup>+</sup>	K <sup>+</sup>	Cl <sup>-</sup>	F <sup>-</sup>	SO <sub>4</sub> <sup>-</sup>	PO <sub>4</sub> <sup>-</sup>	NO <sub>3</sub> <sup>-</sup>	CO <sub>3</sub> <sup>2-</sup>
<b>pH</b>	1.000															
<b>Electrical Conductivity</b>	0.450	1.000														
<b>Total dissolved Solids</b>	0.448	0.342	1.000													
<b>Total Alkalinity</b>	-0.292	-0.353	-0.353	1.000												
<b>Total Hardness</b>	0.193	0.382	0.383	0.228	1.000											
<b>Turbidity</b>	-0.063	0.332	0.334	0.174	0.739	1.000										
<b>Calcium</b>	0.159	0.313	0.315	-0.123	0.468	0.401	1.000									
<b>Magnesium</b>	-0.476	-0.375	-0.373	0.338	-0.034	0.189	0.244	1.000								
<b>Sodium</b>	-0.539	-0.293	-0.291	0.222	0.011	0.223	-0.129	0.408	1.000							
<b>Potassium</b>	-0.016	0.203	0.203	-0.585	-0.025	0.311	0.554	0.206	0.103	1.000						
<b>Chloride</b>	0.084	-0.209	-0.211	0.171	-0.548	-0.676	-0.295	0.034	-0.047	-0.258	1.000					
<b>Fluoride</b>	-0.316	-0.337	-0.338	0.321	-0.047	-0.063	0.369	0.187	-0.138	0.123	0.311	1.000				
<b>Sulphate</b>	-0.123	0.324	0.324	-0.279	-0.160	-0.205	0.394	0.165	0.072	0.394	0.380	0.225	1.000			

<b>Phosphate</b>	0.620	0.107	0.105	-	-	-0.371	-	-0.204	-	-	0.100	-	-	1.000		
<b>Nitrate</b>	-	-	-	0.052	-	-0.316	-	0.604	0.062	0.054	0.463	0.088	0.419	0.193	1.000	
<b>Carbonate</b>	0.449	0.409	0.408	0.130	0.093	0.184	0.083	-0.401	-	-	-	-	-	0.113	0.644	1.000

Table 3.1 shows the correlation between pairs of soil parameters in the spring season. pH with EC, TA, TH,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{Cl}^-$ ,  $\text{PO}_4^-$  and  $\text{NO}_3^-$  there is a positive correlation and with parameters like TDS,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{F}^-$ ,  $\text{SO}_4^-$  and  $\text{CO}_3^{2-}$  it has negative correlation. EC has positive correlation with parameters like TA, TH,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{Cl}^-$ ,  $\text{PO}_4^-$  and  $\text{NO}_3^-$  and has negative correlation with parameters of TDS,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{F}^-$ ,  $\text{SO}_4^-$ , and  $\text{CO}_3^{2-}$ . TDS with TA, TH,  $\text{Ca}^{2+}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{F}^-$ ,  $\text{SO}_4^-$  and  $\text{CO}_3^{2-}$  is a positive correlation, but, with  $\text{Mg}^{2+}$ ,  $\text{Cl}^-$ ,  $\text{PO}_4^-$  and  $\text{NO}_3^-$  it is negative. TA has positive correlation with parameters like TH,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$  and  $\text{K}^+$ , but it has negative correlation with  $\text{Na}^+$ ,  $\text{Cl}^-$ ,  $\text{F}^-$ ,  $\text{SO}_4^-$ ,  $\text{PO}_4^-$ ,  $\text{NO}_3^-$  and  $\text{CO}_3^{2-}$ . Ca has positive correlation with  $\text{K}^+$ ,  $\text{SO}_4^-$ ,  $\text{NO}_3^-$  and  $\text{CO}_3^{2-}$  and has negative correlation with  $\text{Mg}^{2+}$ ,  $\text{Cl}^-$  and  $\text{PO}_4^-$ .  $\text{Mg}^{2+}$  has positive correlation with  $\text{Na}^+$ ,  $\text{Cl}^-$ ,  $\text{SO}_4^-$  and  $\text{PO}_4^-$  and has negative correlation with  $\text{K}^+$ ,  $\text{F}^-$ ,  $\text{NO}_3^-$  and  $\text{CO}_3^{2-}$ .  $\text{Na}^+$  makes positive correlation with  $\text{K}^+$ ,  $\text{Cl}^-$ ,  $\text{F}^-$ ,  $\text{SO}_4^-$ ,  $\text{PO}_4^-$  and  $\text{CO}_3^{2-}$  but, negative correlation with  $\text{NO}_3^-$ . K has positive correlation with  $\text{Cl}^-$ ,  $\text{PO}_4^-$  and  $\text{CO}_3^{2-}$  and has negative correlation with  $\text{F}^-$ ,  $\text{SO}_4^-$  and  $\text{NO}_3^-$ .  $\text{Cl}^-$  has positive correlation with  $\text{SO}_4^-$ ,  $\text{PO}_4^-$ , and  $\text{CO}_3^{2-}$  and has negative correlation with  $\text{F}^-$  and  $\text{NO}_3^-$ .  $\text{F}^-$  has positive correlation with  $\text{SO}_4^-$ ,  $\text{PO}_4^-$ ,  $\text{NO}_3^-$  and  $\text{CO}_3^{2-}$ .  $\text{SO}_4^-$  has positive correlation with  $\text{PO}_4^-$  and  $\text{CO}_3^{2-}$  and has negative correlation with  $\text{NO}_3^-$ .  $\text{PO}_4^-$  has positive correlation with  $\text{CO}_3^{2-}$  and has negative correlation with  $\text{NO}_3^-$ .  $\text{NO}_3^-$  has negative correlation with  $\text{CO}_3^{2-}$ .

#### 4. Linear Regression analysis of spring season soil samples

In linear regression for electrical conductivity and parameters composition of spring season of soil samples were analysed. We have observed that the “t” value for spring season is greater than 0.03, which indicates that there is a high relationship among the electrical conductivity and parameters composition in spring season of soil samples.

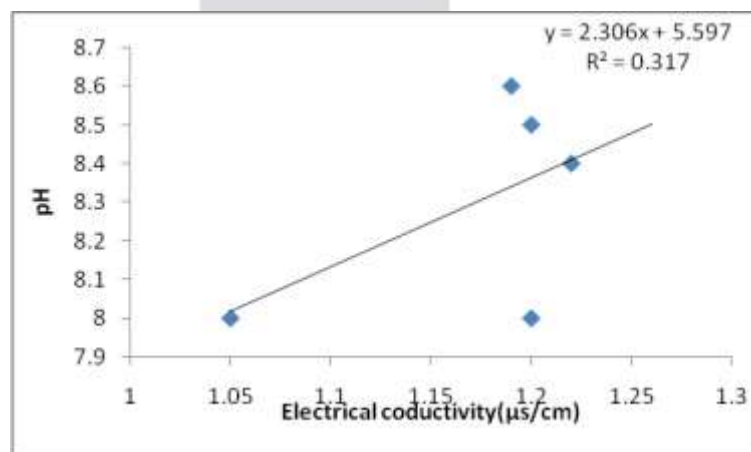
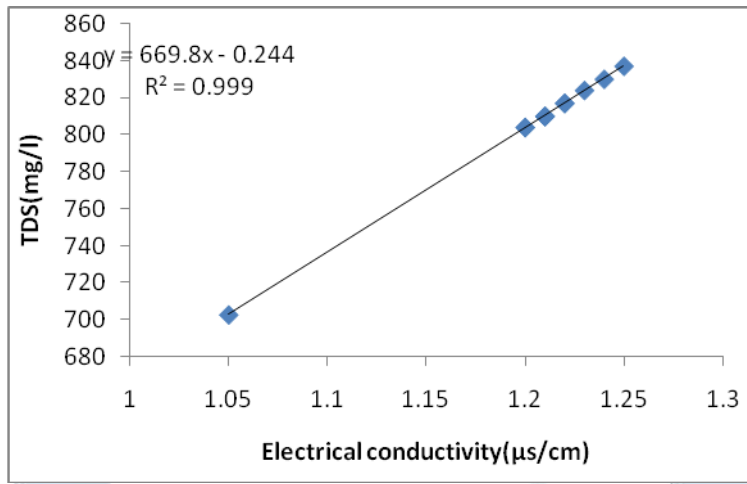
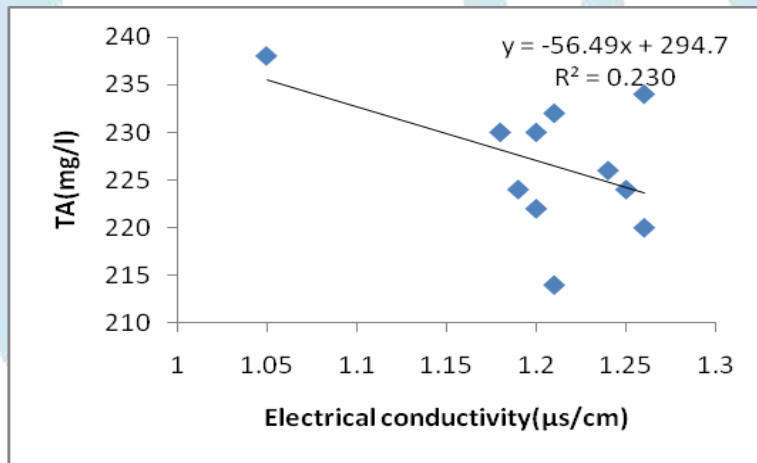


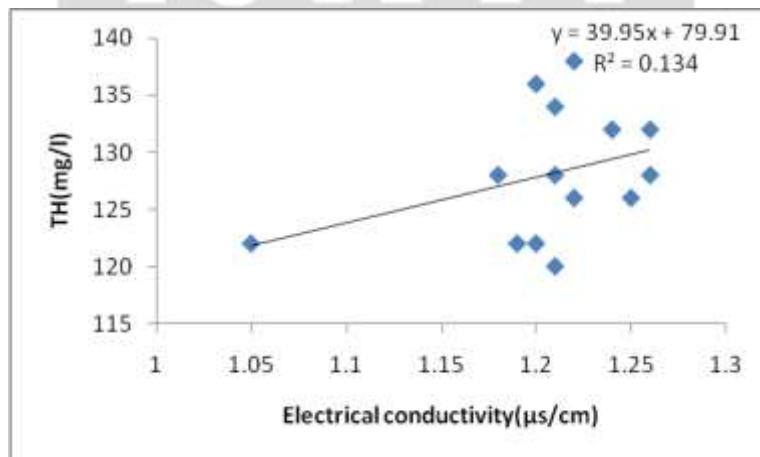
Fig 4 (i) Electrical conductivity vs pH



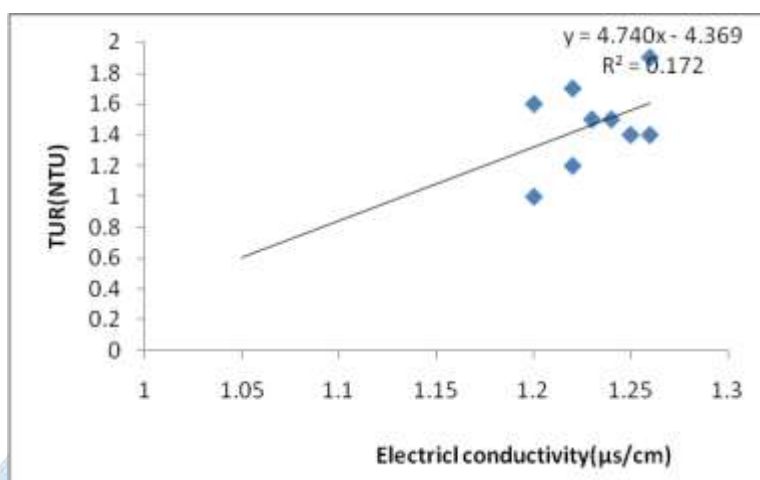
**Fig 4 (ii) Electrical conductivity vs Total dissolved solids**



**Fig 4 (iii) Electrical conductivity vs Total alkalinity**



**Fig 4 (iv) Electrical conductivity vs Total hardness**



**Fig 4 (v) Electrical conductivity vs Turbidity**

Figure 4 (i-v) Linear regression analysis of spring season soil samples shows the X axis is Electrical conductivity with Y axis pH, Total dissolved solids, Total Alkalinity, Total Hardness and Turbidity. The linear regression values is

$R^2 = 0.317, 0.999, 0.230, 0.134$  and  $0.172$ . The spring season value is

$R^2 = 0.999$ (TDS) is significant level.

#### 4. Conclusion

Soil samples collected from spring seasons in various site. In order to investigate the Physico-chemical analysis in each sample and to correlate the results with the extent of pollutions, the following spectral and physico-chemical studies are successfully carried out. All the above mentioned parameters and the minerals are within the permissible limit. The presence of minerals is within the permissible limit, which increases cultivation.

#### 5. References

- Devi S and Premkumar R, (2012), "Physicochemical analysis of groundwater samples near industrial area, Cuddalore District, Tamilnadu, India", *International journal of chemtech research*, **4(1)**, 29-34.
- Inbanila T and V.Arutchelvan, (2012), "Study on groundwater quality in and around sipcot industrial complex, area cuddalore district, tamilnadu", *International Journal of Chem Tech Research*, **4(1)**, 29-34.
- Selvaraju R and Anitha B, (2021), "Hydrochemistry of groundwater in and Around the Industrial area of cuddalore District, Tamilnadu", *Journal of Emerging Technologies and Innovative Research (JETIR)*, **8(12)**, 774-780.
- Selvaraju R and Anitha B, (2022), "Correlation and Regression Analysis to Determine the Percentage existence of Physico-chemical Parameters on Electrical Conductivity in the Industrial Area Drinkingwater around Cuddalore Old town, Tamilnadu", *International Journal of Current Research and Review (IJCR)*, **14(4)**, 7-14.
- Selvaraju R and Mahalakshmi V, (2024), "Physico- Chemical, Correlation and Linear Regression Analysis of Small Industry Area Soil Samples (Winter Season)", *International Journal of Research in Academic World(IJRAW)*, **3(3)**, 115-118.
- Sajil Kumar P. J and E. J. James, (2013), "Physicochemical parameters and their sources in groundwater in the Thirupathur region, Tamil Nadu, South India", *Applied water Science*, **3**, 219-228.

- Thilagavathi T, Kanchana S, Banumathi P, Hemalatha G, Vanniarajan C, Sundar M and Ilamaran M, (2015), “Physico-chemical and functional characteristics of selected millets and pulses”, *Indian Journal of Science and Technology*, **8(7)**, 147-155.
- Udhayakumar R, Manivannan P, Raghu K and Vaideki S, (2016), “Assessment of physico-chemical characteristics of water in Tamilnadu”, *Ecotoxicology and environmental safety*, **134**, 474-477.
- Umamageswari T.S.R, Sarala Thambavani D and Liviu M, (2019), “Hydrogeochemical processes in the groundwater environment of Batlagundu block, Dindigul district, Tamil Nadu: conventional graphical and multivariate statistical approach”, *Applied Water Science*, **9(1)**, 1-15.
- Umarani P, Ramu A and Kumar V, (2019), “Hydrochemical and statistical evaluation of groundwater quality in coastal aquifers in Tamil Nadu, India”, *Environmental Earth Sciences*, **78(15)**, 1-14.

